LOYOLA COLLEGE (AUTONOMOUS), CHENNAI - 600 034



B.Sc. DEGREE EXAMINATION - **CHEMISTRY**

FIFTH SEMESTER - NOVEMBER 2018

CH 5402- POLYMER CHEMISTRY

Date: 03-11-2018 Dept. No. Max. : 100 Marks

Time: 09:00-12:00

Part A

Answer ALL questions.

 $(10 \times 2 = 20)$

- 1. What is monomer functionality?
- 2. Provide an example for natural and synthetic polymer.
- 3. Define cohesive energy.
- 4. Give any two commonly used inhibitors in free radical polymerization.
- 5. What is compounding?
- 6. What is vulcanization of rubber?
- 7. Expand the following terms: (i) HDPE (ii) PMMA.
- 8. What are elastomers?
- 9. Differentiate between thermo and thermo setting plastics.
- 10. What are the characteristics of photostabilizers?

Part B

Answer any EIGHT questions.

 $(8 \times 5 = 40)$

- 11. Illustrate and explain the tacticity of polymers.
- 12. Mention the characteristic features of monomers and polymers.
- 13. Describe the mechanism of anionic polymerization of styrene.
- 14. Compare addition and condensation polymerizations.
- 15. Write a short note on conducting polymers.
- 16. How is epoxy resin synthesized? Mention its uses.
- 17. What are polymer additives? Mention the advantages of fillers.
- 18. Describe the free radical mechanism of polymerization of propylene.
- 19. Write a note on interfacial polymerization.
- 20. Classify and explain the polymers based on their thermal behavior.
- 21. Give an account on thermosetting plastics with suitable examples.
- 22. Write a short note on bulk polymerization technique.

Part C

Answer any FOUR questions.

 $(4 \times 10 = 40)$

- 23. Discuss the primary and secondary bond forces in polymers.
- 24. Derive the formulae for number and weight average molecular weight of polymers.
- 25. Discuss the mechanism of Ziegler-Natta polymerization with a suitable example.
- 26. Write an account on various types of polymer degradation process.
- 27. Describe the calendaring and die casting polymerization process.
- 28. Write a note on conducting polymers and fibre forming polymers.
